

isotherms Studies. Saturation magnetization M_s values were calculated through the interception of M versus $1/H$ curve with the vertical axis.) The excitation source was 532 nm radiation from a Nd:YVO₄ laser (frequency-doubled) and the laser power on the sample was 5 mW. Magnetic properties were examined using a Quantum Design PPMS DynaCool-9 System with a vibrating sample magnetometer (VSM) option, with a scanning speed of 0.025 deg/s and a step time of 10 s. The Fourier transformed infrared (FTIR) technique was conducted using a Nicolet Nexus 670 FTIR infrared spectrometer within a range from 4000 to 400 cm^{-1} with a resolution of 4 cm^{-1} in a KBr wafer. For isotherms studies, 0.01 g of either ZnFe₂O₄, NiFe₂O₄, or CoFe₂O₄ ferrite samples was mixed with 10 mL of aqueous solution at pH = 2 for different concentrations of Pb(II) (10, 40, 80, and 100 mg/L). Langmuir and Freundlich isotherm models are commonly used by various researchers to describe the equilibrium of heavy metal ions between solid and solution phases [35]. The Langmuir equation is expressed as $q_e = \frac{q_m K_L C_e}{1 + K_L C_e}$, where q_e is the amount adsorbed at equilibrium (mg/g), C_e is the equilibrium concentration (mg/L), K_L is the Langmuir constant related to the affinity of binding site (L/mg), and q_m is the maximum amount of solute adsorbed (mg/g). The Langmuir isotherm model considers that the binding sites are homogeneously distributed on the adsorbent surface and the adsorption takes place at specific homogeneous sites within the adsorbent. Characterization Techniques: radiation ($\lambda = 10$