

N-Dodecane, also known as dodecane or normal dodecane, is a saturated hydrocarbon with the chemical formula $C_{12}H_{26}$. Conclusion: N-Dodecane is a versatile and important chemical in various industrial applications, primarily valued for its stability, non-reactivity, and solvent properties. Physical State and Appearance: State: Liquid at room temperature Color: Colorless Odor: Odorless Melting and Boiling Points: Melting Point: $-9.6^{\circ}C$ ($14.7^{\circ}F$) Boiling Point: $216.2^{\circ}C$ ($421.2^{\circ}F$) Density: Density: Approximately 0.75 g/cm^3 at $20^{\circ}C$ Solubility: Water Solubility: Insoluble in water Solubility in Organic Solvents: Soluble in organic solvents such as alcohols, ethers, and benzene Viscosity: Viscosity: Low viscosity, which increases slightly with decreasing temperature Flash Point: Flash Point: Approximately $74^{\circ}C$ ($165^{\circ}F$) Applications: Industrial Solvent: Due to its non-polarity and chemical inertness, n-dodecane is commonly used as a solvent in organic reactions and in the formulation of coatings, paints, and varnishes. Chemical and Physical Properties: Molecular Structure: Chemical Formula: $C_{12}H_{26}$ Molecular Weight: 170.33 g/mol Structure: Linear chain of 12 carbon atoms, each bonded to sufficient hydrogen atoms to satisfy the valency of carbon ($C-C-C-C-C-C-C-C-C-C-C-C$). Fuel Research: N-Dodecane is used as a reference fuel in combustion research and studies related to internal combustion engines due to its similarity to diesel fuels.