

Isotherms Studies. Saturation magnetization M_s values were calculated through the interception of M versus $1/H$ curve with the vertical axis.) The excitation source was 532 nm radiation from a Nd:YVO₄ laser (frequency-doubled) and the laser power on the sample was 5 mW. Magnetic properties were examined using a Quantum Design PPMS DynaCool-9 System with a vibrating sample magnetometer (VSM) option, to 70 K, with a scanning speed of 0.025 deg/s and a step time of 10 s. The Fourier transformed infrared (FTIR) technique was conducted using a Nicolet Nexus 670 FTIR infrared spectrometer within a range from 4000 to 400 cm⁻¹ with a resolution of 4 cm⁻¹ in a KBr wafer. For isotherms studies, 0.01 g of either ZnFe₂O₄, NiFe₂O₄, or CoFe₂O₄ ferrite samples was mixed with 10 mL of aqueous solution at pH = 2 for different concentrations of Pb(II) (10, 40, 80, and 100 mg/L). Langmuir and Freundlich isotherm models are commonly used by various researchers to describe the equilibrium of heavy metal ions between solid and solution phases [35]. The Langmuir equation is expressed as $q_e = \frac{q_m K_L C_e}{1 + K_L C_e}$, where q_e is the amount adsorbed at equilibrium (mg/g), C_e is the equilibrium concentration (mg/L), K_L is the Langmuir constant related to the affinity of binding site (L/mg), and q_m is the maximum amount of solute adsorbed (mg/g). The Langmuir isotherm model considers that the binding sites are homogeneously distributed on the adsorbent surface and the adsorption takes place at specific homogeneous sites within the adsorbent.

Characterization Techniques. radiation ($\lambda = 10$