

The adsorption of MB onto pulverized Marula seed husks was feasible, endothermic and spontaneous from the thermodynamics data evaluated. An optimum pH of 10 was established for the adsorption process. Although the equilibrium data fitted well into the Langmuir and Freundlich isotherms at various temperatures, the Langmuir isotherm best described the data. Marula seed husk can be