Schiff bases have important role in the development of [coordination & medicine] chemistry as they readily form new complexes with metals [8]. These compounds are played in the field of bioinorganic chemistry and various aspects of organometallic compounds[9]. They are formed by condensation of a primary amine (RNH2) and carbonyl compound. The (– HC=N–) group is particularly suited for binding to metal ions via the N atom lone pair (–N:) and when contain one or more donor atoms in addition to (– C=N–) group they act as[polydentate ligands or macrocycles]. Schiff base and its complexes containing azomethine group (–HC=N–) as shown in scheme(1– 1).